



1. Background: Engineering Biodegradability into Responsive Materials

- Thermo-responsive polymers exhibiting a Lower Critical Solution Temperature (LCST) have been exploited for drug release¹ and hyperthermia-triggered cellular uptake.^{2,3}
- For many *in vivo* applications, these materials should be degradable.⁴
- Triggering an "isothermal" LCST (changing solubility without altering temperature) may also be desirable in some cases (e.g. to avoid protein denaturation).⁵⁻⁷
- Here, we introduce reduction-sensitive disulphide bonds into thermo-responsive poly(*N*-isopropylacrylamide), pNIPAM, using a novel polycondensation-type procedure.
- Importantly, this material selectively degrades at intracellular (mM) glutathione (GSH) concentrations but remains stable at extracellular (μ M) concentrations.⁸
- We combine this with the known molecular weight dependent LCST behaviour of pNIPAM⁹ to trigger an "isothermal" transition (Fig. 1).

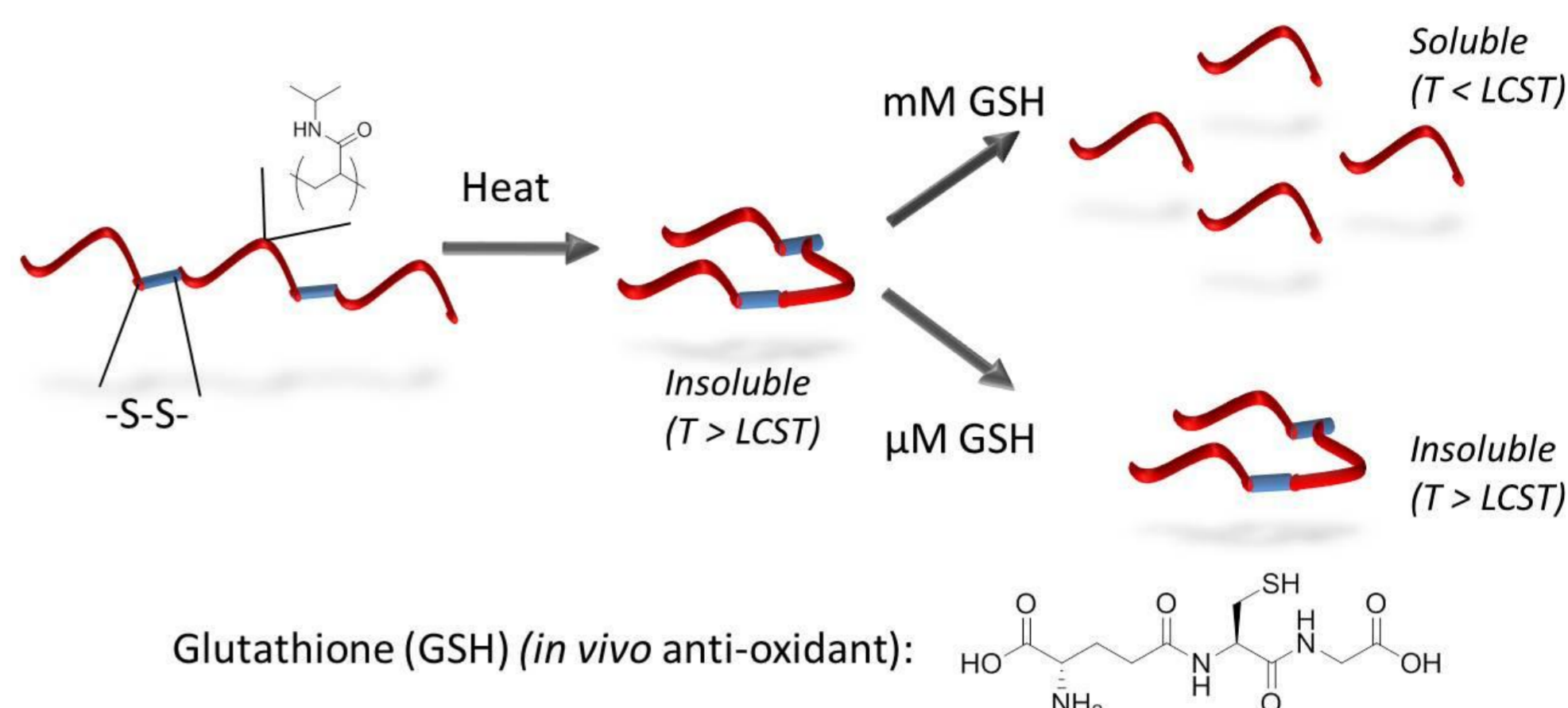


Figure 1. Schematic illustration of study aims.

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2. Preparing Disulphide-Linked Polymers

- Disulphides are formed by thiol exchange reactions with pyridyl disulphide (PDS).¹⁰
- Dithioester and trithiocarbonate RAFT CTAs (Fig. 2A) were used to polymerise a variety of PDS-containing (meth)acrylates and acrylamides.
- MALDI-ToF of pNIPAM prepared with CTA I quantified the presence of both end-groups on each chain (Fig. 2B).
- Disulphide links were introduced using a polycondensation-type procedure; aminolysis of thiocarbonyl end-group followed by *in situ* thiol-disulphide exchange (Fig. 2C).

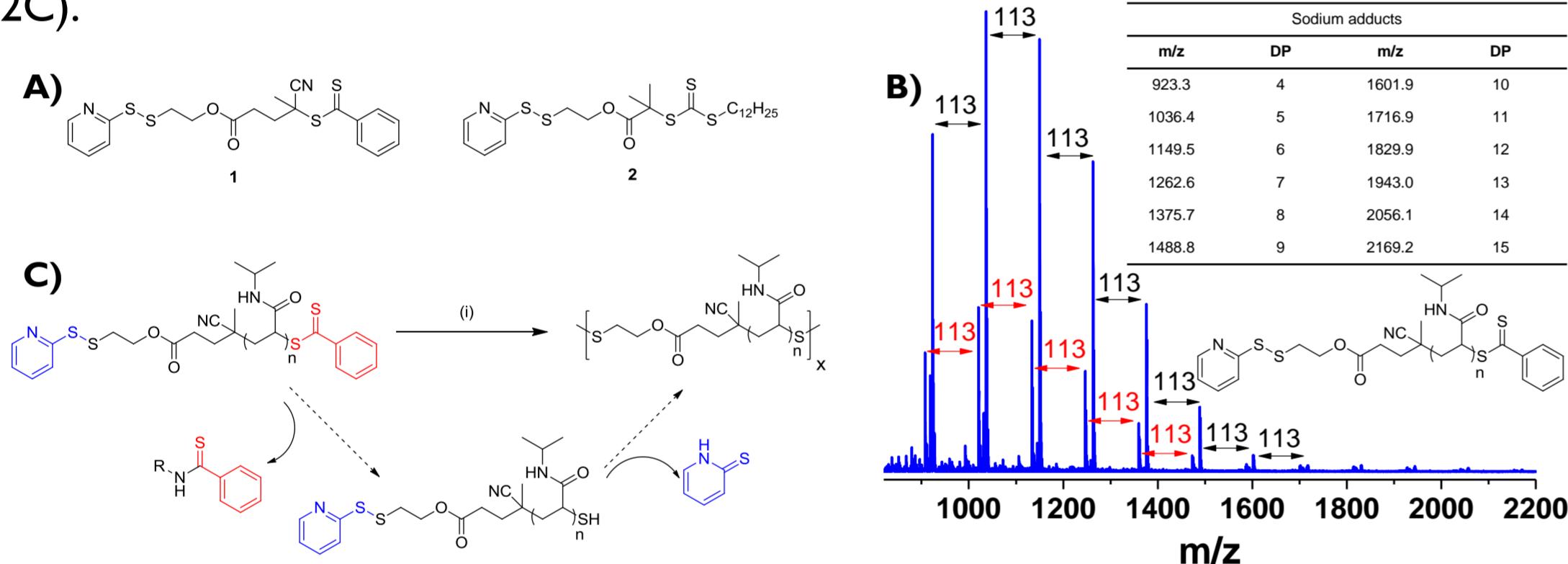


Figure 2. A) CTAs used to prepare PDS-functionalised polymers. B) MALDI-ToF spectrum of pNIPAM prepared with CTA I. Peaks correspond to *n*-NIPAM units, both end-groups and a lithium (red) or sodium (black, inset) ion. C) Polycondensation of telechelic pNIPAM: (i) Ethanolamine (1 eq.); triethylamine (2 eq.); THF; N₂; 25 °C; 24 h.

3. Polycondensation Scope

- Disulphide linkages were successfully incorporated into pNIPAM (Fig. 3A). An increase in molecular weight from 1755 to 34000 g.mol⁻¹ was observed.
- Degradability was confirmed by reduction with tributyl phosphine.
- A variety of monomers were tested (Fig. 3B). This synthetic methodology was compatible with all except non-linear poly(ethylene glycol) analogues.

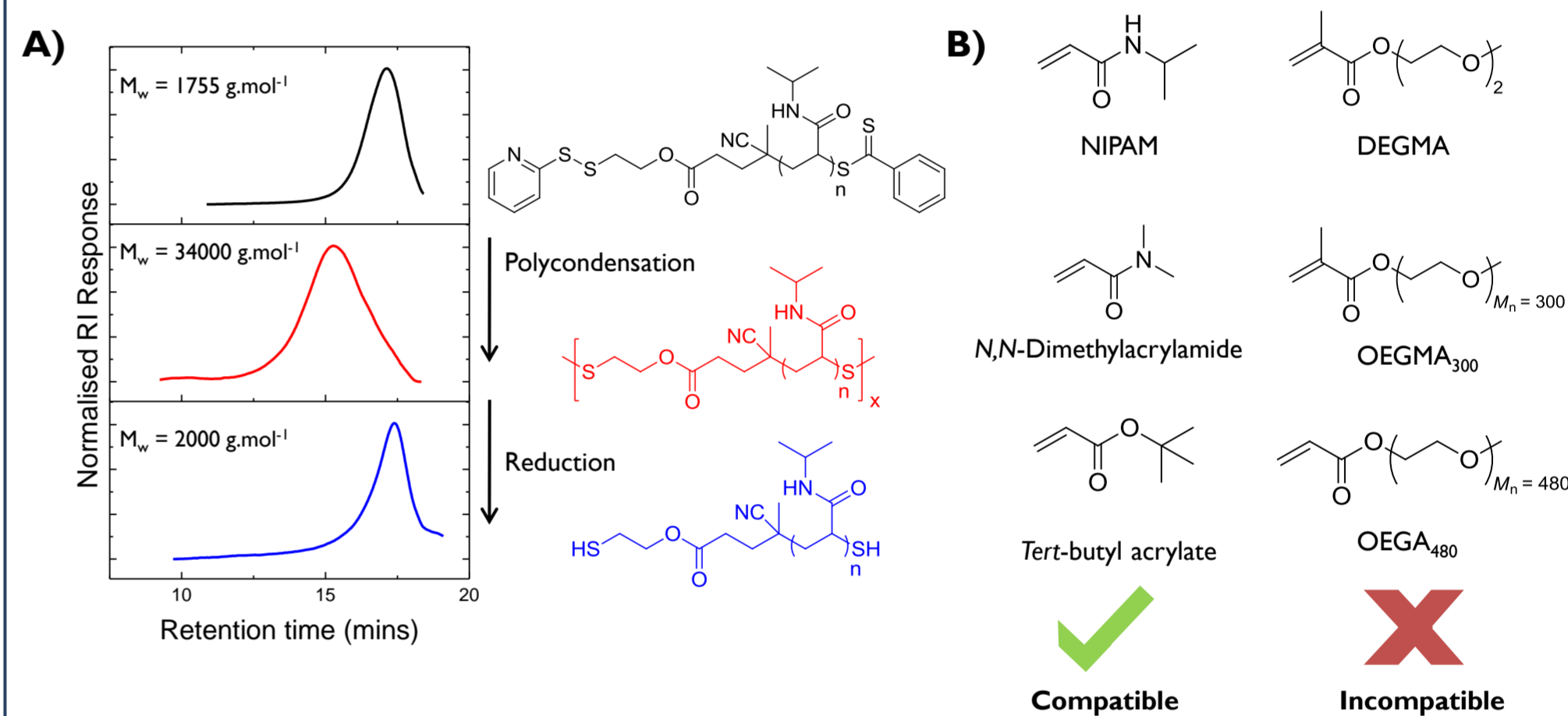


Figure 3. A) SEC analysis demonstrating successful pNIPAM polycondensation procedure. Reduction was achieved using tributyl phosphine. B) Chemical structure of other monomers tested in this study.

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4. Triggering an "Isothermal" Transition

- Disulphide-linked materials (ss-pNIPAM) from a variety of pNIPAM starting molecular weights were prepared.
- The cloud point (CP, measurable property of LCST) of pNIPAM and ss-pNIPAM (before and after incubation in mM GSH solution) was recorded (Fig. 4).
- A CP shift from 30 °C (ss-pNIPAM) to 39 °C (ss-pNIPAM, 1 mM GSH) was observed.

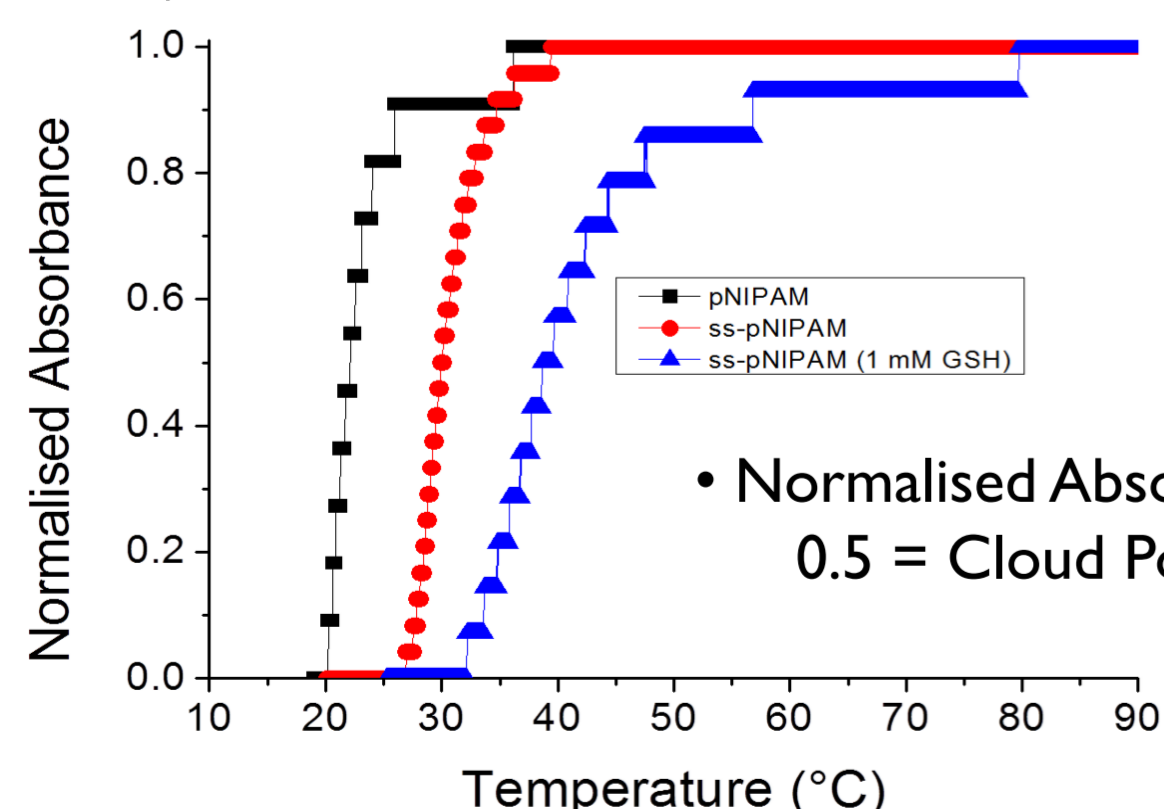


Figure 4. Cloud point determination of pNIPAM, ss-pNIPAM and ss-pNIPAM following mM GSH incubation.

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- This shift was used to promote an isothermal transition (Fig. 5):
- Incubation of ss-pNIPAM in water above its CP led to its precipitation;
- Treatment with mM GSH solution caused rapid re-solubilisation;
- No re-solubilisation was observed at μ M GSH concentrations.

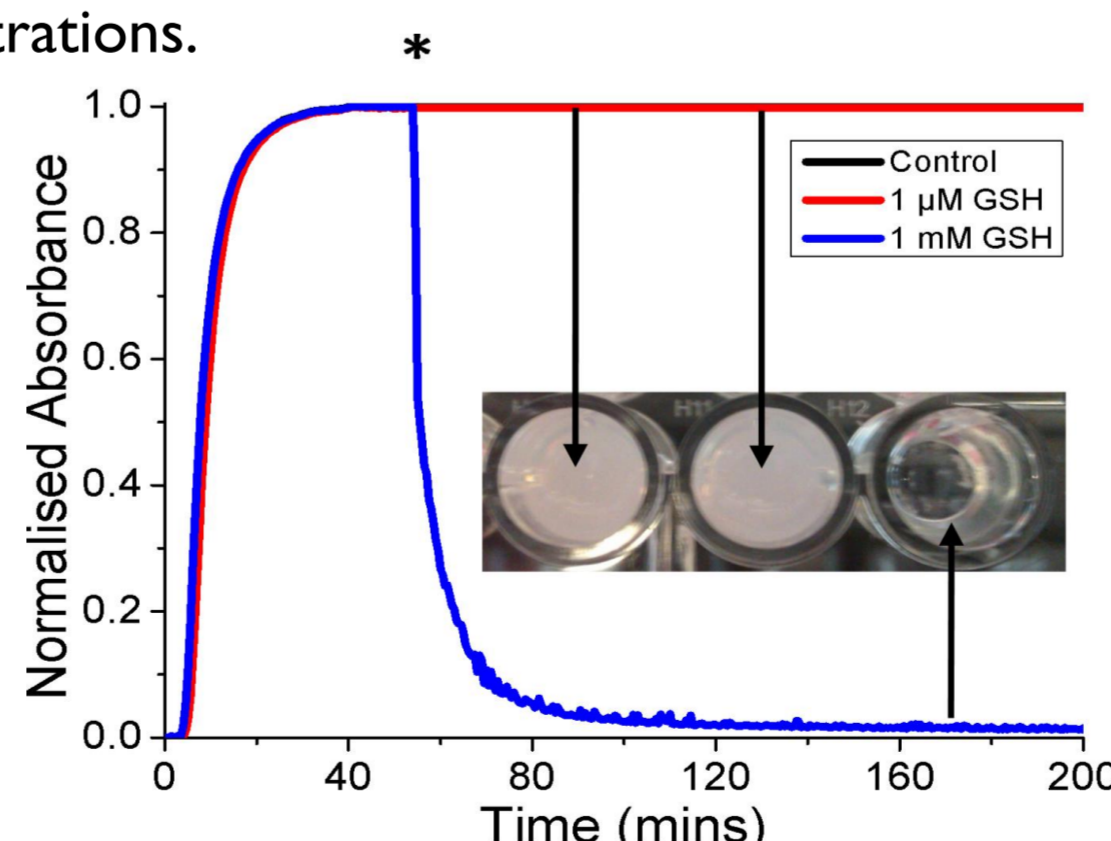


Figure 5. Isothermal turbidimetry data for ss-pNIPAM. Temperature = 37 °C; * = GSH added. Inset: photo at end of experiment, left = control; centre = 1 μ M GSH; right = 1 mM GSH.

5. Summary

- Pyridyl disulphide-functionalised pNIPAM was prepared in a controlled manner using both dithiobenzoate and trithiocarbonate chain transfer agents.
- Disulphides were incorporated into the polymeric structure via a polycondensation-type reaction between a terminal thiol and pyridyl disulphide group.
- Selective degradability of disulphide linkages to intracellular glutathione concentration was demonstrated.
- This was combined with the known molecular weight dependent LCST of pNIPAM to produce an isothermal transition.
- Future work will use these materials to trigger mammalian cellular uptake.

6. References

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