

Low Temperature Heat Recovery and Distribution Network Technologies

Dr Linwei Wang

Research Associate

Centre for Renewable Energy Systems Technology (CREST)

Loughborough University

L.Wang2@lboro.ac.uk



Loughborough
University



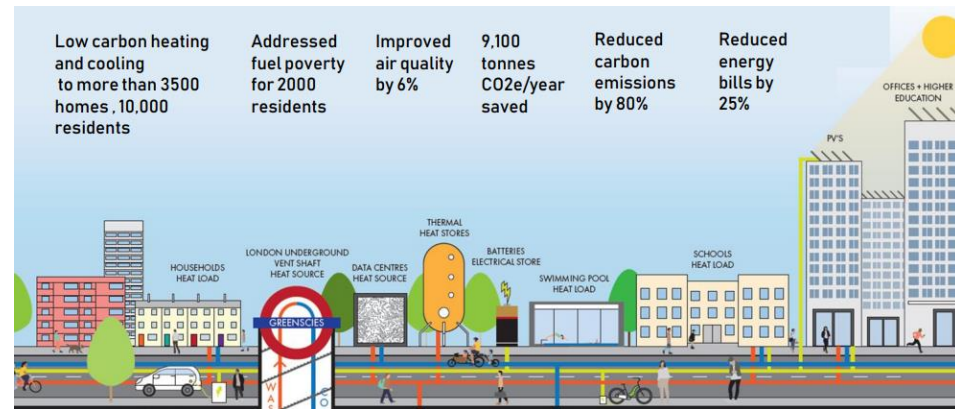
London South Bank
University

Research Challenge

LoT-NET research focuses on the integration of low temperature (LT) networks with heat pumps and thermal storage technologies to maximise waste and ambient heat utilisation in low or zero-carbon heating and cooling solutions.

Research Challenge 2: To advance performance of novel thermal storage, heat distribution and capture systems.

- **WP2.1** Distribution medium or method: Thermochemical systems;
- **WP2.2** Storage: New composite thermochemical heat storage materials will be developed.



Thermal Energy Storage

Sensible Heat Storage

- Temperature changes
- Water, Oil, Rocks etc.
- $Q = V\rho C_p\Delta T$

- Low cost
- Reliable
- Low storage density
- Large volumes
- Heat losses

Latent Heat Storage

- Phase changes
- Salts, Paraffins, Eutectic etc.
- $Q = mh_m$

- Medium storage density
- Small volumes
- Corrosion
- Low heat transfer
- Heat Losses

Thermochemical Storage

- Chemical reactions
- Solid/Liquid-Gas, Gas-Gas
- $Q = Xn_B\Delta H_r$

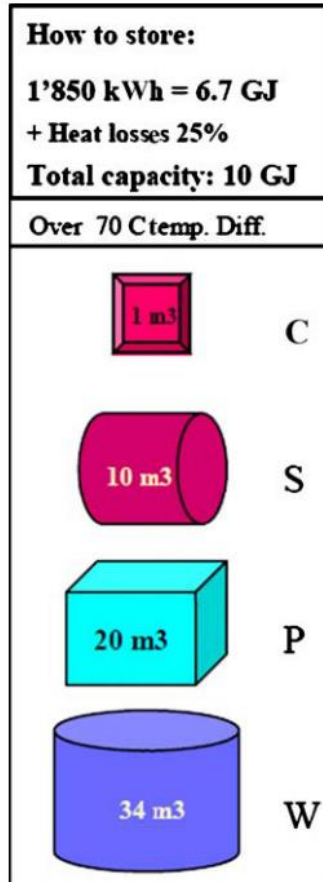
- Medium/Long term storage
- High storage density
- Technical complexity
- High costs

Thermal Energy Storage

Thermochemical Storage

Latent Heat Storage

Sensible Heat Storage



Energy density
In MJ/m³

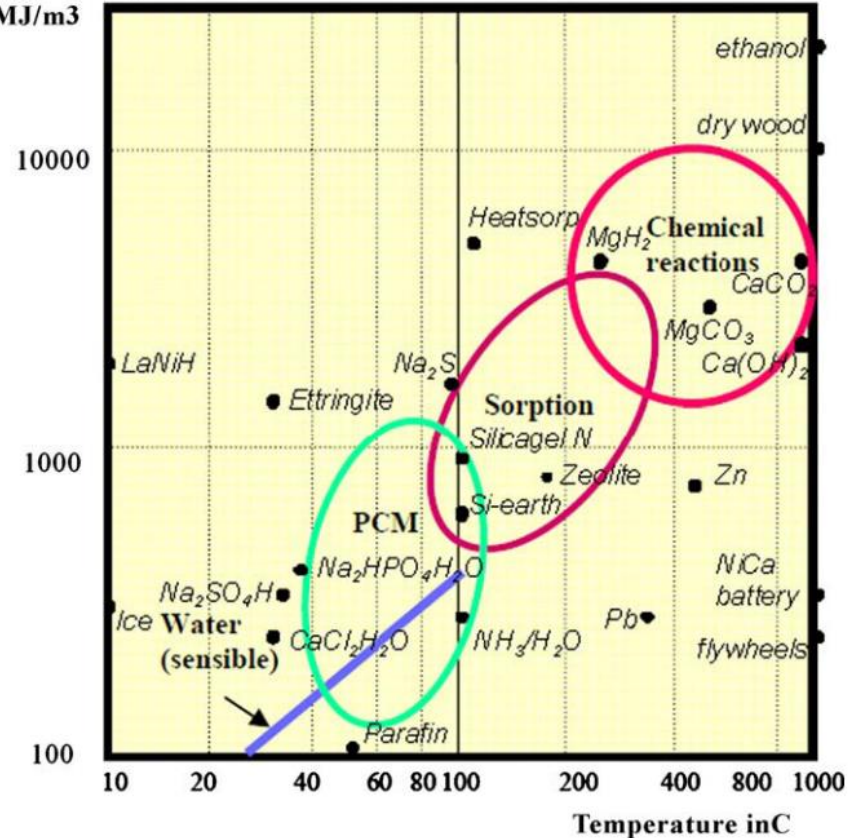


Fig.1. Energy density of high energy storage methods. [1]

[1] N'Tsoukpoe KE, Liu H, Pierres NL, Luo LG. A review on long-term sorption solar energy storage. Renewable and Sustainable Energy Reviews, 2009, 13, 2385-2396.

Thermochemical Heat Storage

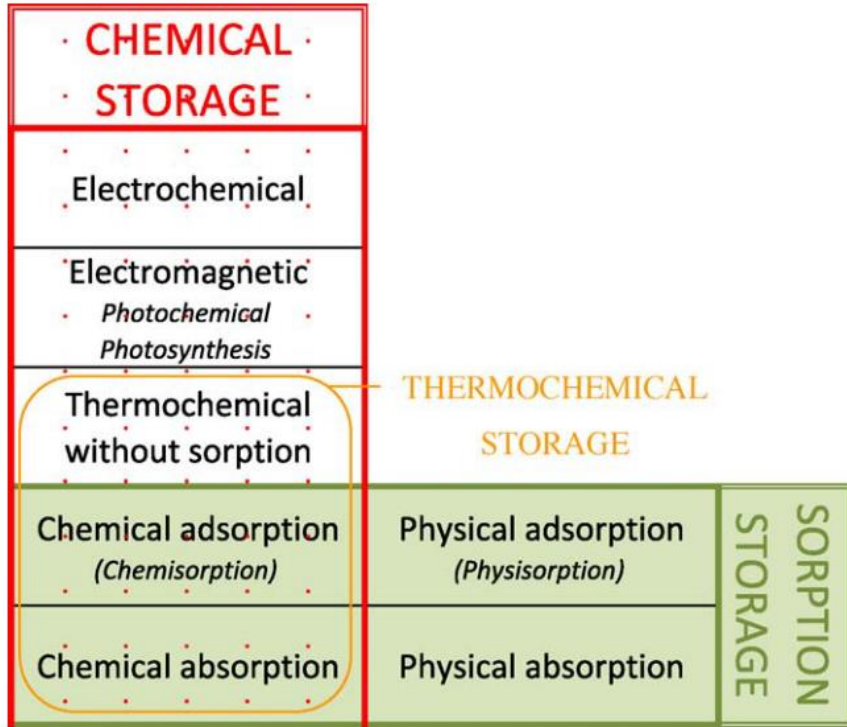
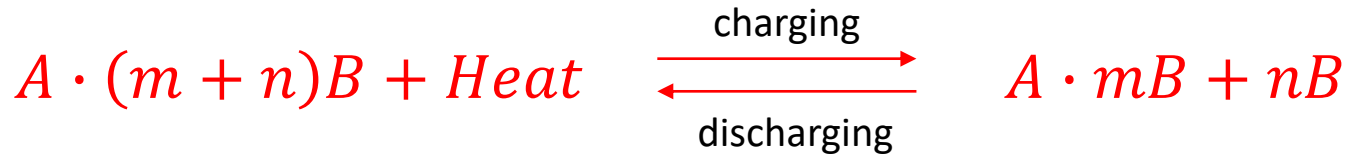


Fig.2. Chemical storage and sorption storage classification. [1]

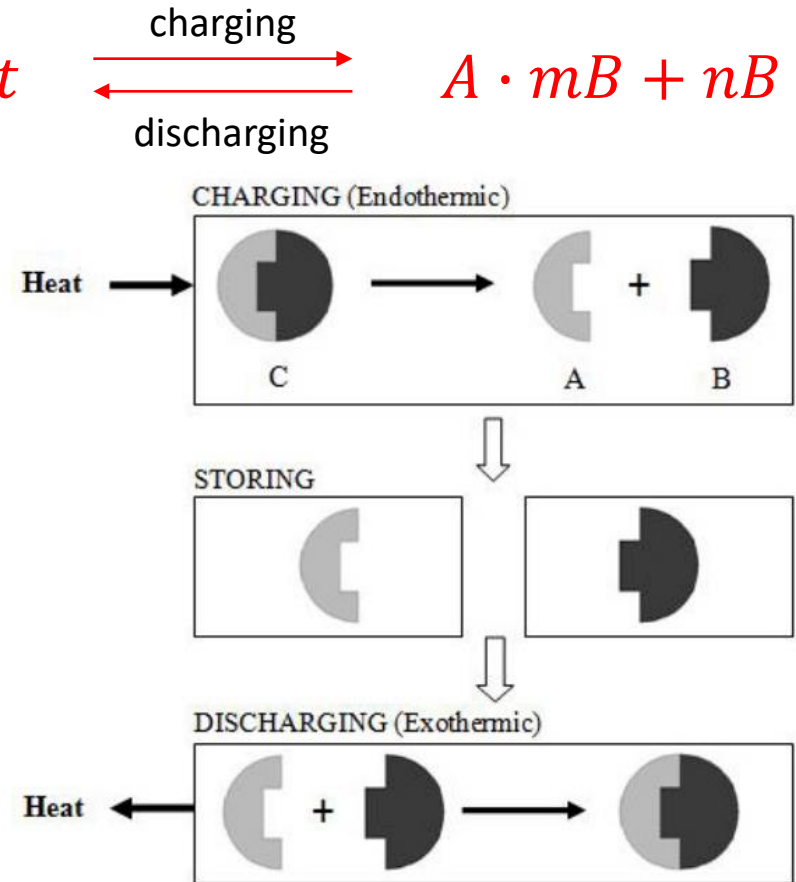


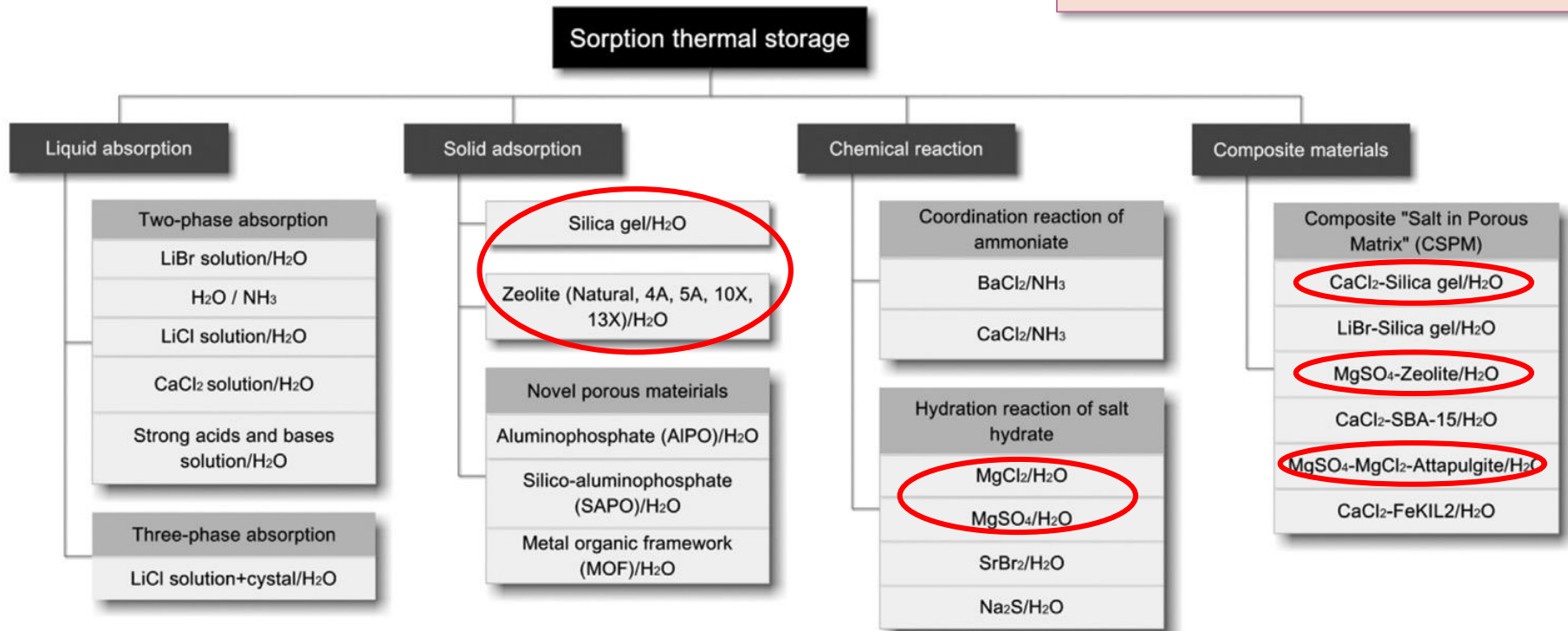
Fig.3. Charging, storing and discharging processes involved in a thermochemical energy storage cycle. [2]

[2] Abedin AH, Rosen MA. A critical review of thermochemical energy storage systems. The Open Renewable Energy Journal, 2011, 4, 42-46.

Thermochemical Heat Storage

Sorption thermal storage classification [3]

Waste Heat Source: 120°C



[3] Yu N, Wang RZ, Wang LW. Sorption thermal storage for solar energy. Progress in Energy and Combustion Science, 2013, 39, 489-514.

Example: $MgCl_2 \cdot 6H_2O$

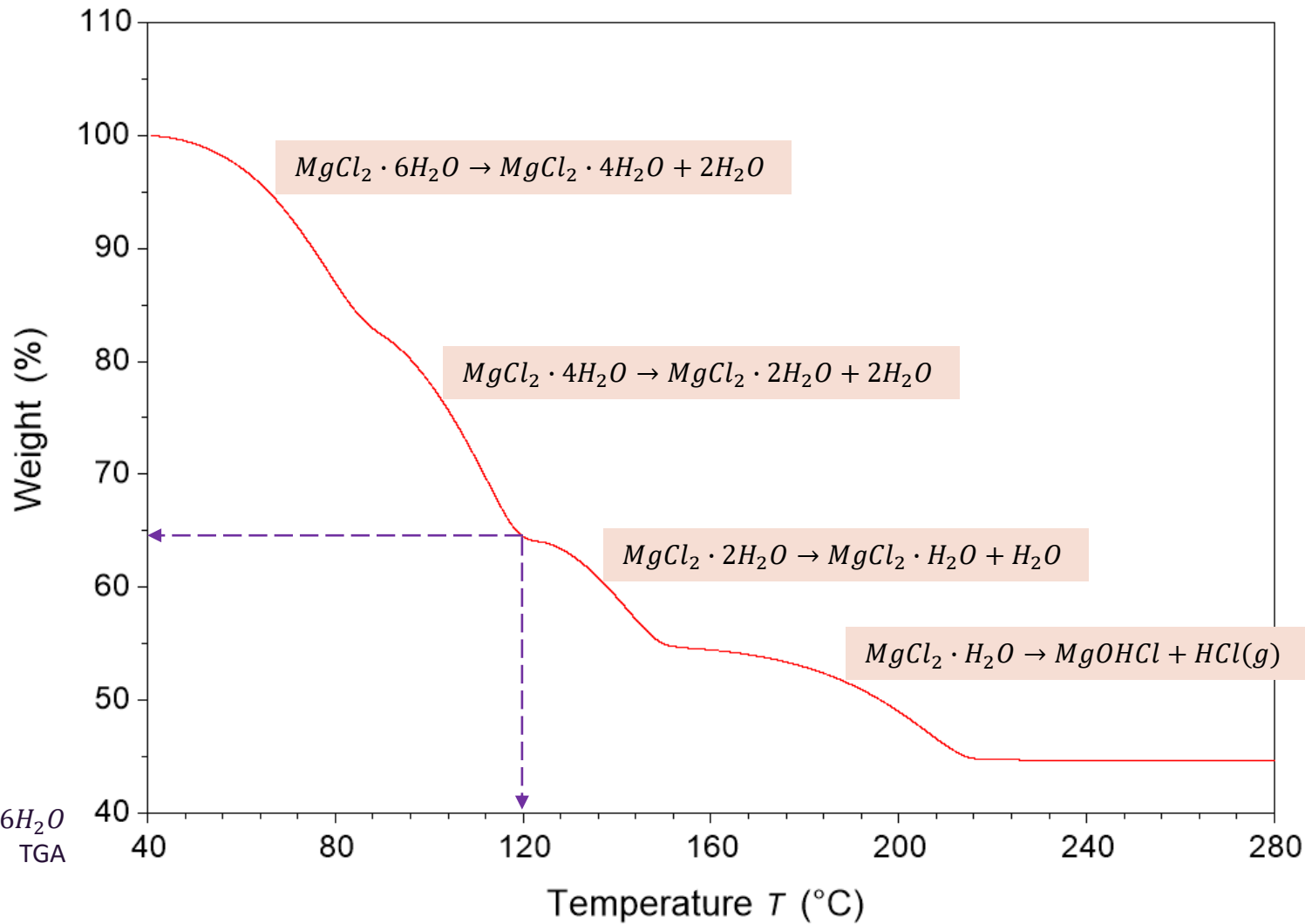


Fig.4. Weight loss of $MgCl_2 \cdot 6H_2O$ versus temperature using a TGA with heating rate $2^\circ\text{C}/\text{min}$.

Example: $MgCl_2 \cdot 6H_2O$

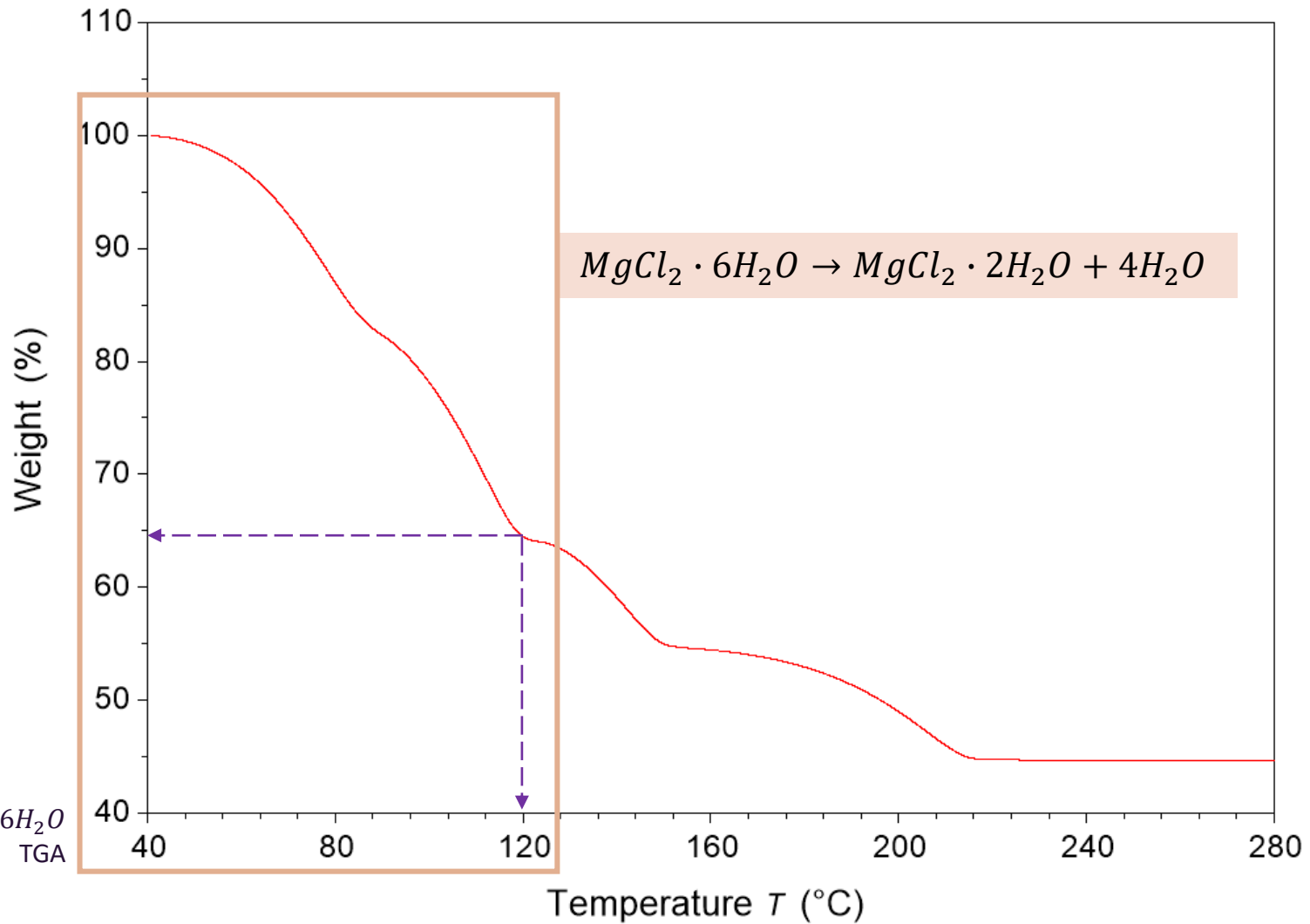


Fig.4. Weight loss of $MgCl_2 \cdot 6H_2O$ versus temperature using a TGA with heating rate $2^\circ\text{C}/\text{min}$.

Example: $MgCl_2 \cdot 6H_2O$

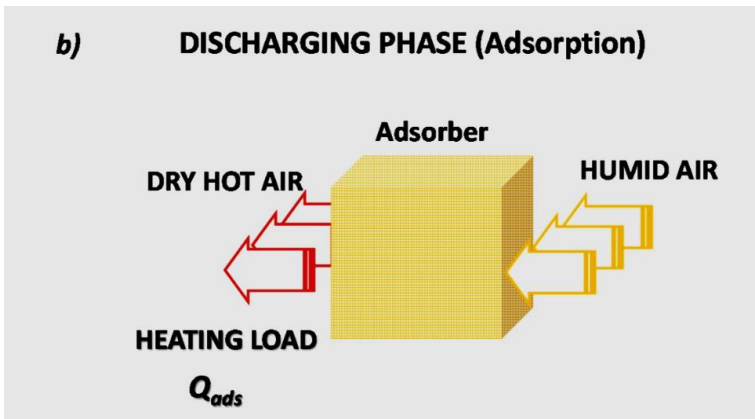
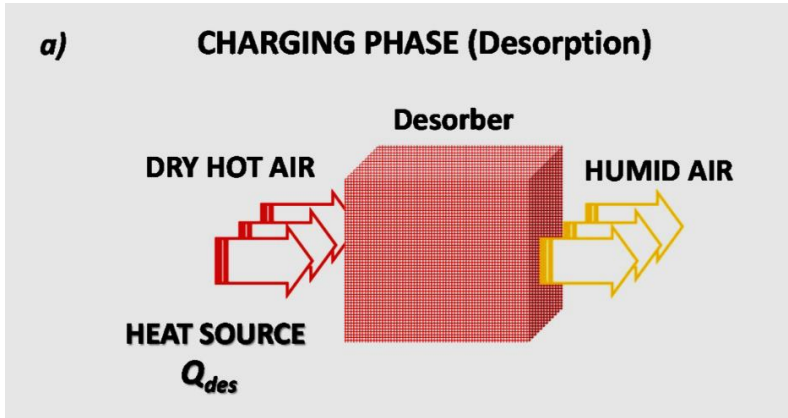


Fig.5. An adsorption heat storage cycle. [4]

[4] Vasta S, Brancato V, Rosa DL, Palomba V, Restuccia G, Sapienza A, Frazzica A. Adsorption Heat Storage: State-of-the-Art and Future Perspectives. *Nanomaterials*, 2018, 8, 522.

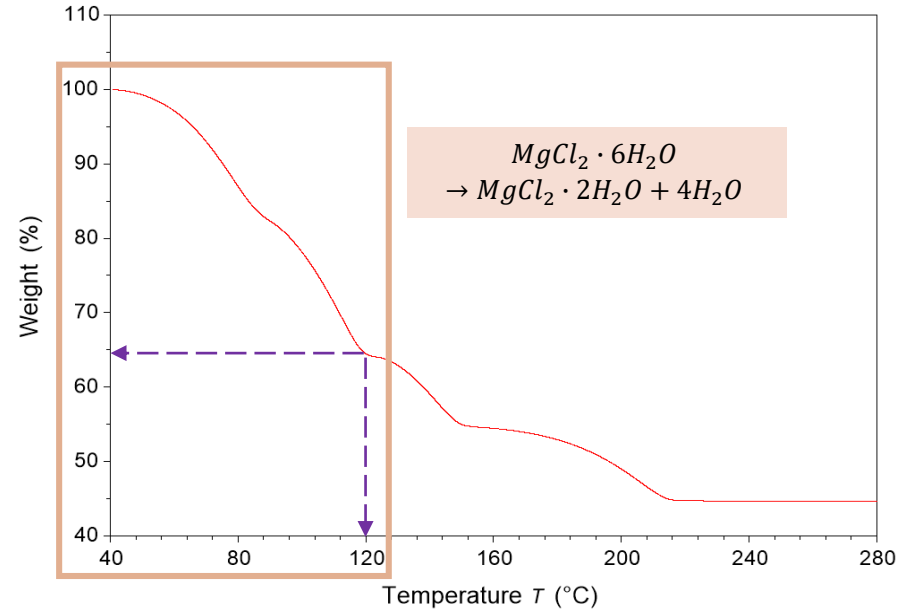


Fig.4. Weight loss of $MgCl_2 \cdot 6H_2O$ versus temperature using a TGA with heating rate $2^{\circ}C/min$.

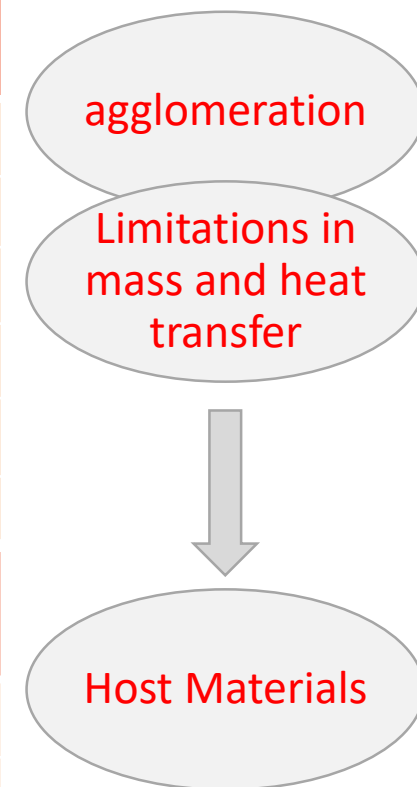


Potential Materials for TCES

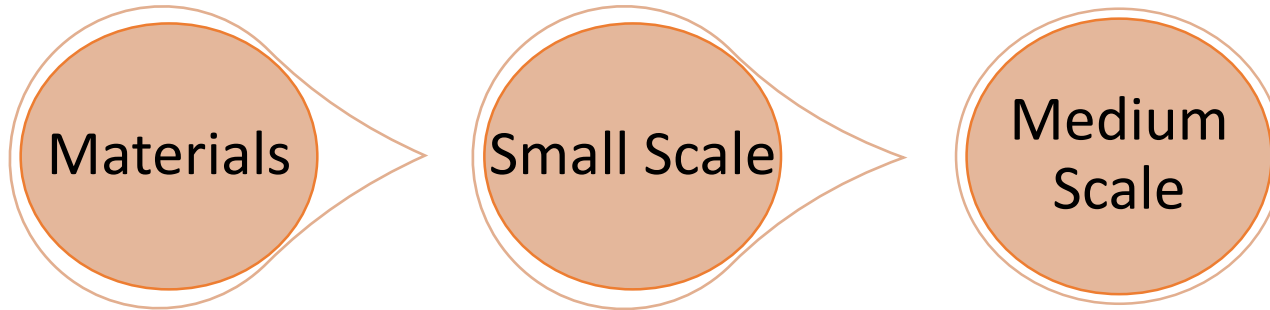
Table 1: Details of Potential Materials

Materials	Charging Temperature	Storage Density (Wh/kg)	Supplier	Price
$MgCl_2$	130°C	477	Fisher Scientific	1KG/£99.20
$MgSO_4$	150°C	442	Fisher Scientific	1KG/£106.00
$CaCl_2$	100°C	623	Fisher Scientific	1KG/£46.00
$SrCl_2$	100°C		Fisher Scientific	
AIPO-18	95°C	243		
SAPO-34	95°C	203		

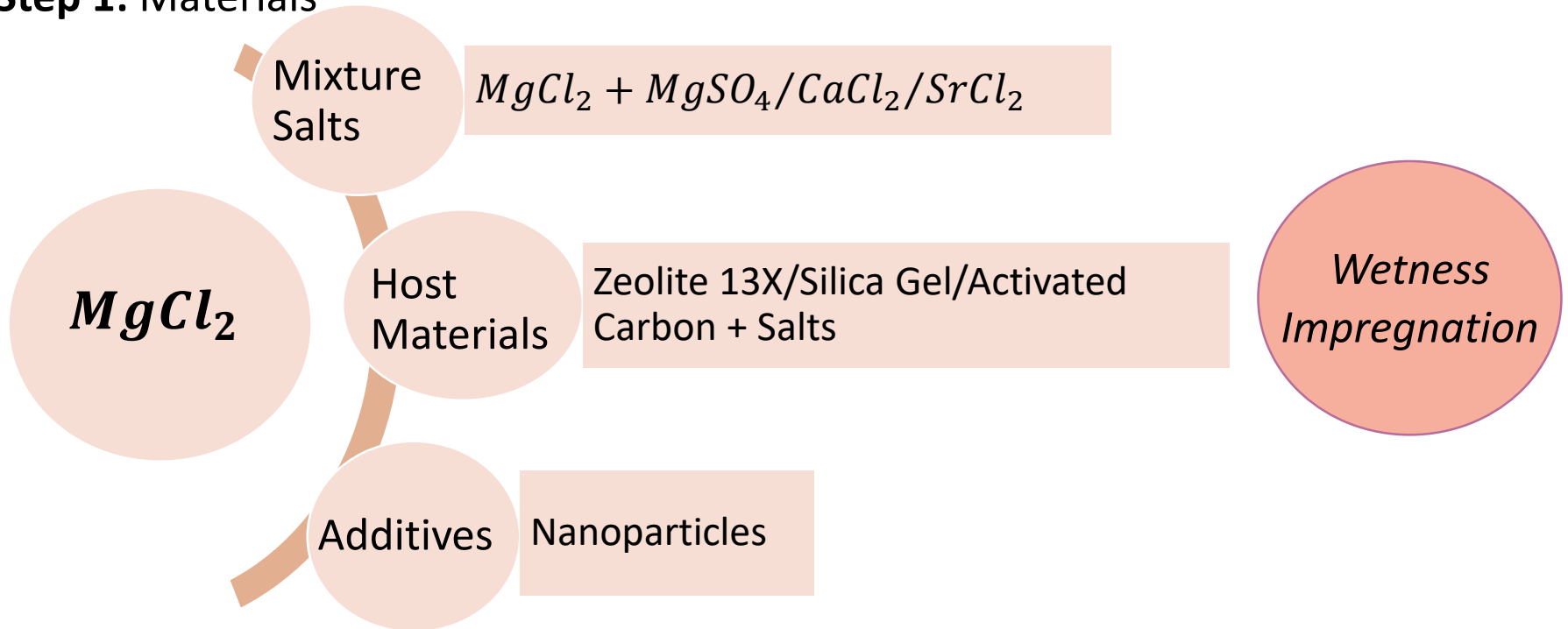
Materials	Charging Temperature	Storage Density (Wh/kg)	Supplier	Price
Zeolite 13X			Fisher Scientific	1KG/£38.08
Silica Gel			Fisher Scientific	1KG/£58.20
Activated Carbon			Sigma Aldrich	1KG/£51.10



Research Plan



Step 1: Materials



Research Plan

Materials

Small Scale

Medium Scale

Step 1: Materials

Mixture Salts

$MgCl_2$

Host Materials

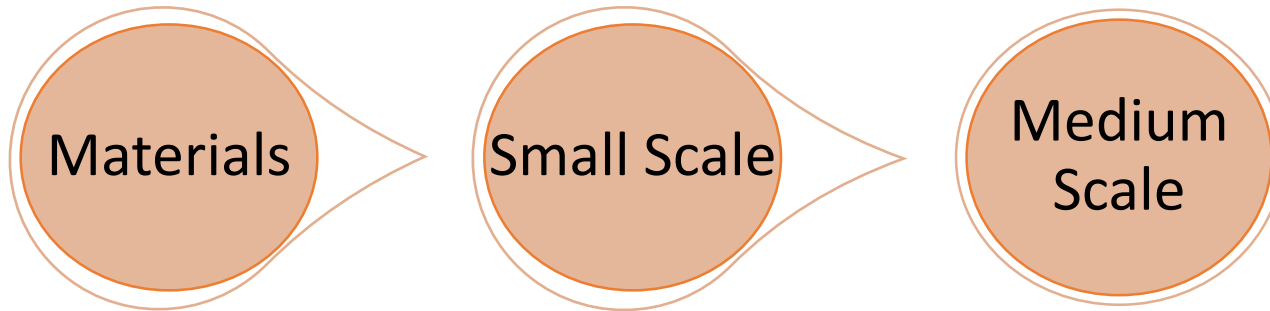
Additives



Thermophysical Properties

Theoretical Storage Density

Research Plan



Step 2: Hydration and dehydration cycle tests in small scale (~10mg).



Dehydration
(charging)

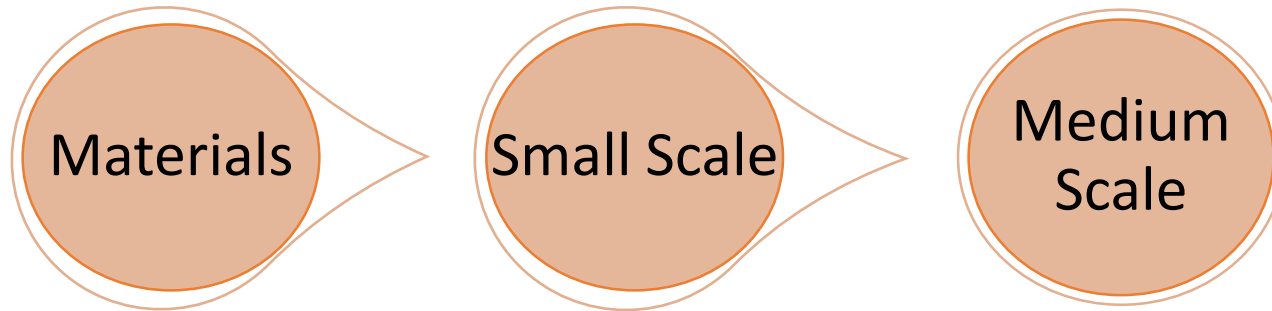
Hydration
(discharge)

Cycle Tests



- ✓ Cycle Stability
- ✓ System Energy Storage Density $W = \frac{Q_{discharge}}{Volume}$
- ✓ Energy Storage Efficiency $\eta = \frac{Q_{discharge}}{Q_{charge}}$

Research Plan



Step 3: Experimental Reactor and System Design in medium scale (~50g).

✓ **System:** Open/Closed

✓ **Reactor:** Particles

Beads

Pellets

Honeycombs

Things To Do

- Making materials for hydration/dehydration cycle tests;
 - ❖ Lab training (TGA/DSC done);
- Conducting thermophysical analysis of materials;
- Conducting hydration/dehydration cycle tests;
- Designing rig for medium scale experiments.



Low Temperature Heat Recovery and Distribution Network Technologies

THANK YOU!

Dr Linwei Wang
Research Associate
CREST, Loughborough University
L.Wang2@lboro.ac.uk

