



**REDESIGN THERMOCHEMICAL ENERGY
STORAGE REACTOR SYSTEM FOR THE
APPLICATION OF WATER AND RESIDUAL
SALTS FROM MINING ACTIVITY**

WARWICK
THE UNIVERSITY OF WARWICK

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Warwick Sustainable Thermal Energy Technologies

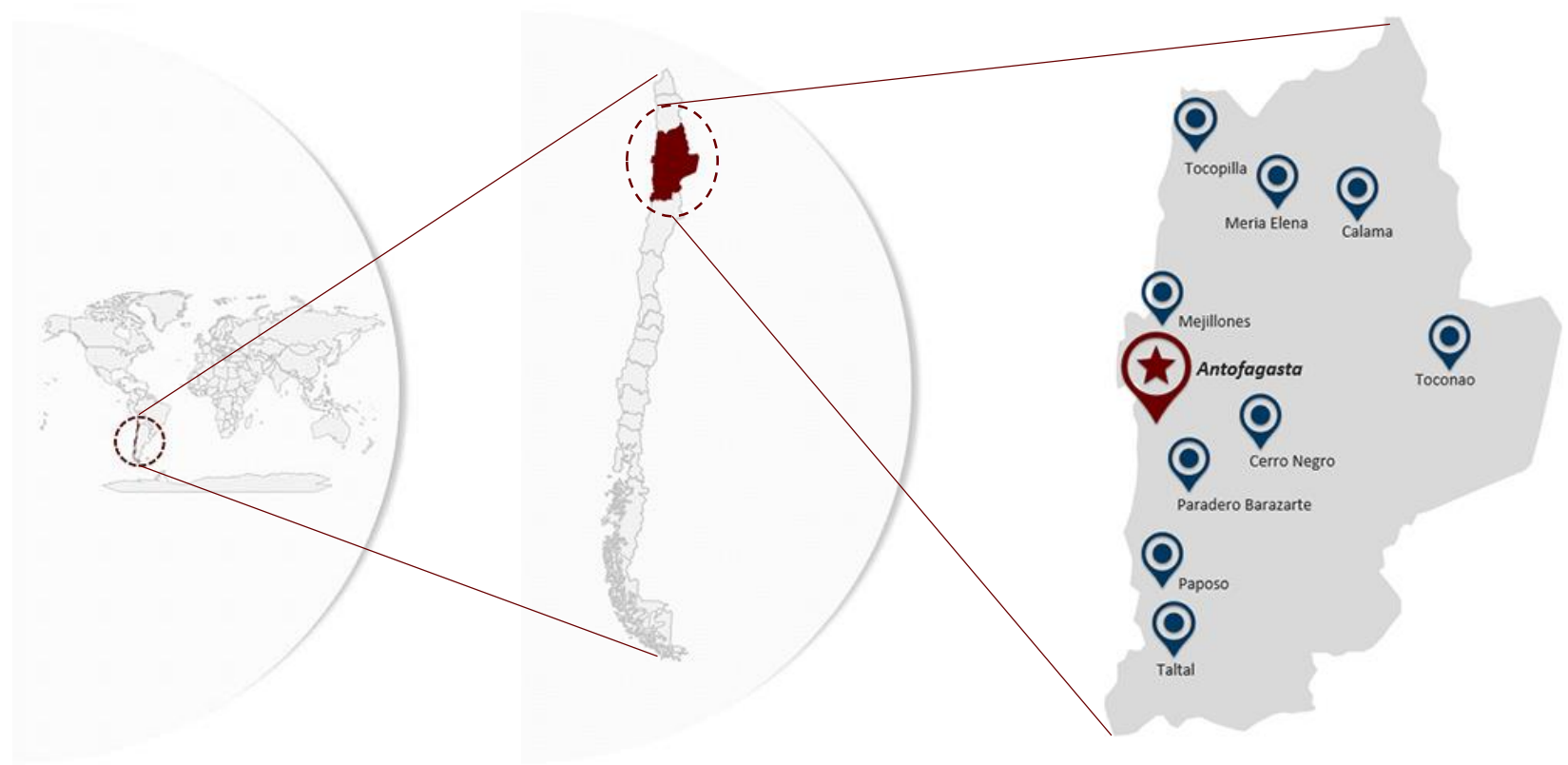


Overview

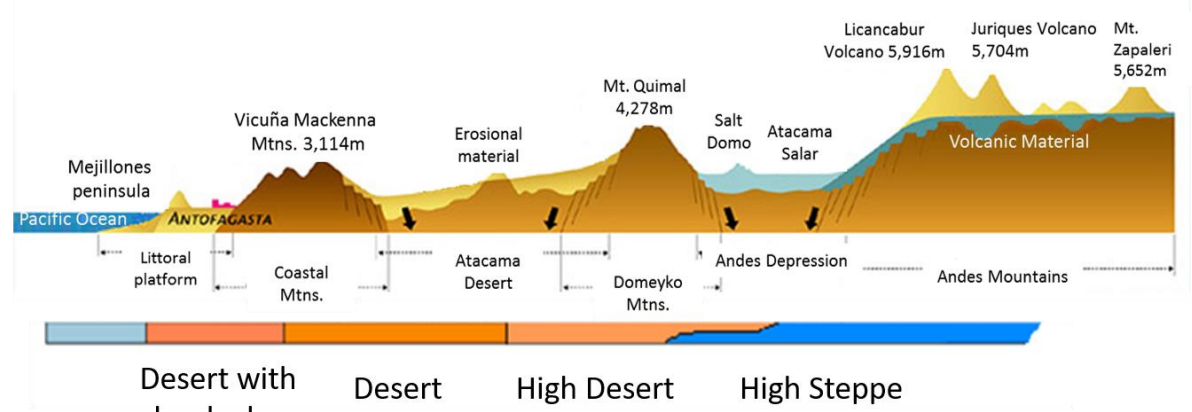
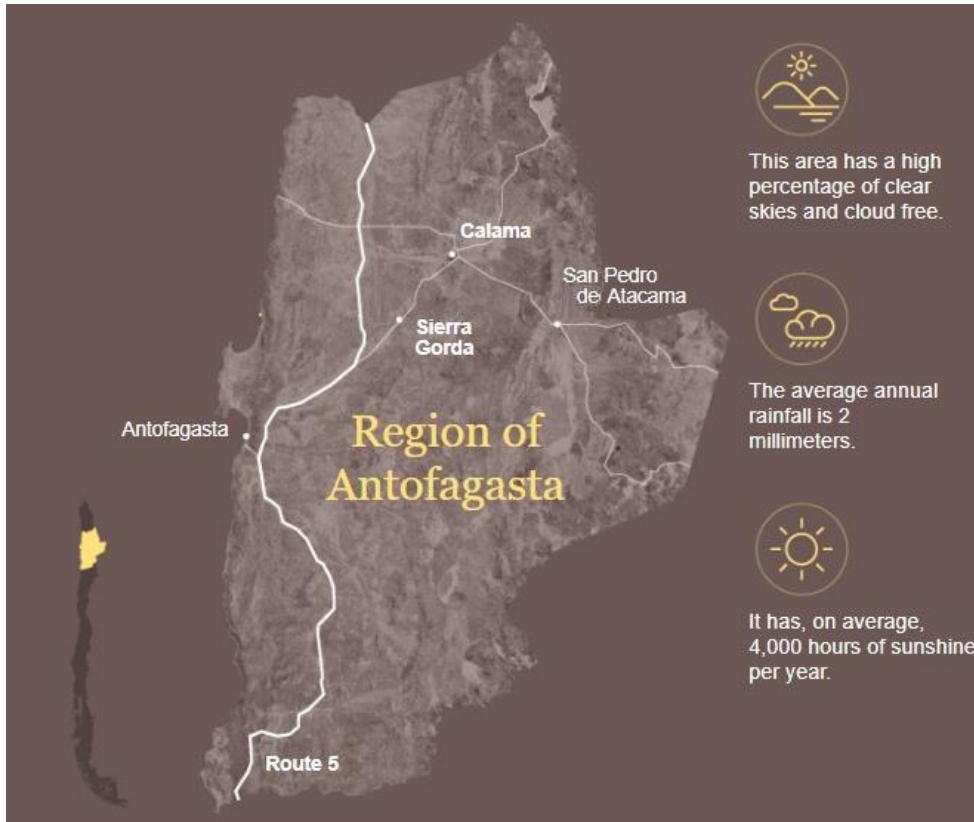
- **Background**
- PhD research
- Research proposal STET group

Chile, Antofagasta Region

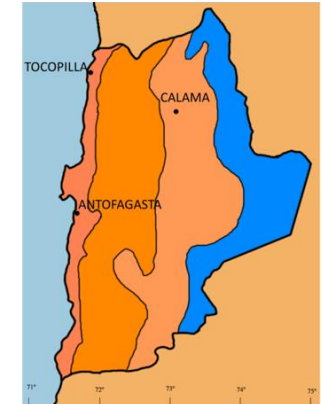
Chile, located in South America, fills a narrow 2,880 miles strip between The Andes and the Pacific Ocean.



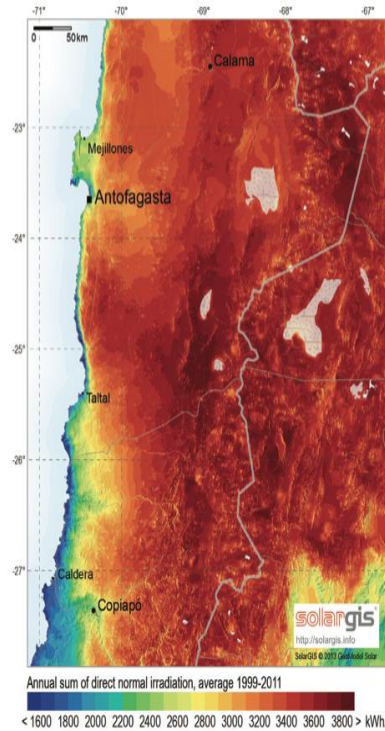
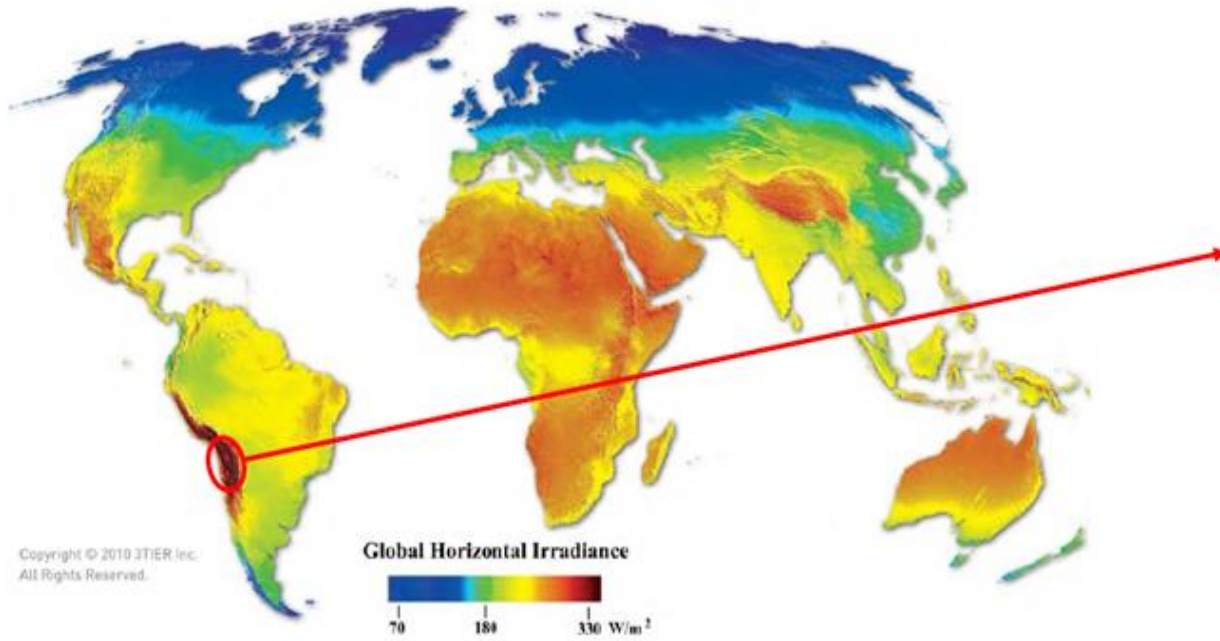
Climates of the Antofagasta Region



Climate	Height	Annual T mean			Rainfall
		min	max	δ	
Cloudy Desert	> 0 m	13 °C	22 °C	± 7 °C	1.7 mm
Normal Desert	> 1000 m	8 °C	21 °C	± 25 °C	3.1 mm
High Desert	> 2000 m	5 °C	24 °C	± 20 °C	5.6 mm
High Steppe	> 3000 m	5 °C	24 °C	± 20 °C	68 mm



Antofagasta in the Global Solar Resource Map



High solar radiation > 2500 kWh/m² (GHI) per year

Temperature fluctuations in the interior of the desert between -15°C to 50°C.

Mining, the main activity of the region



Several processes for mineral extraction

Residual salts

Bischofite ($\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$)

Lithium carnallite ($\text{LiMgCl}_3 \cdot 6\text{H}_2\text{O}$)

Potassium carnallite ($\text{KMgCl}_3 \cdot 6\text{H}_2\text{O}$)

Kainite ($\text{KMg}(\text{SO}_4)\text{Cl} \cdot 3\text{H}_2\text{O}$)

Astrakanite ($\text{Na}_2\text{Mg}(\text{SO}_4)_2 \cdot 4\text{H}_2\text{O}$)

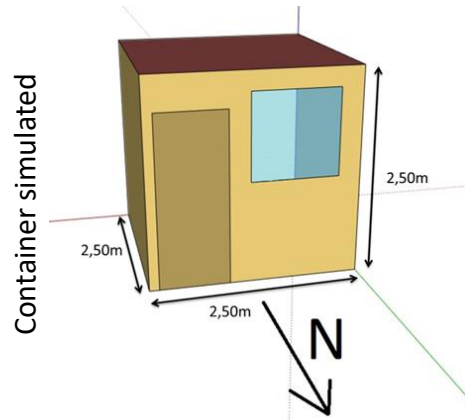
Lithium nitrate ($\text{LiNO}_3 \cdot 3\text{H}_2\text{O}$)



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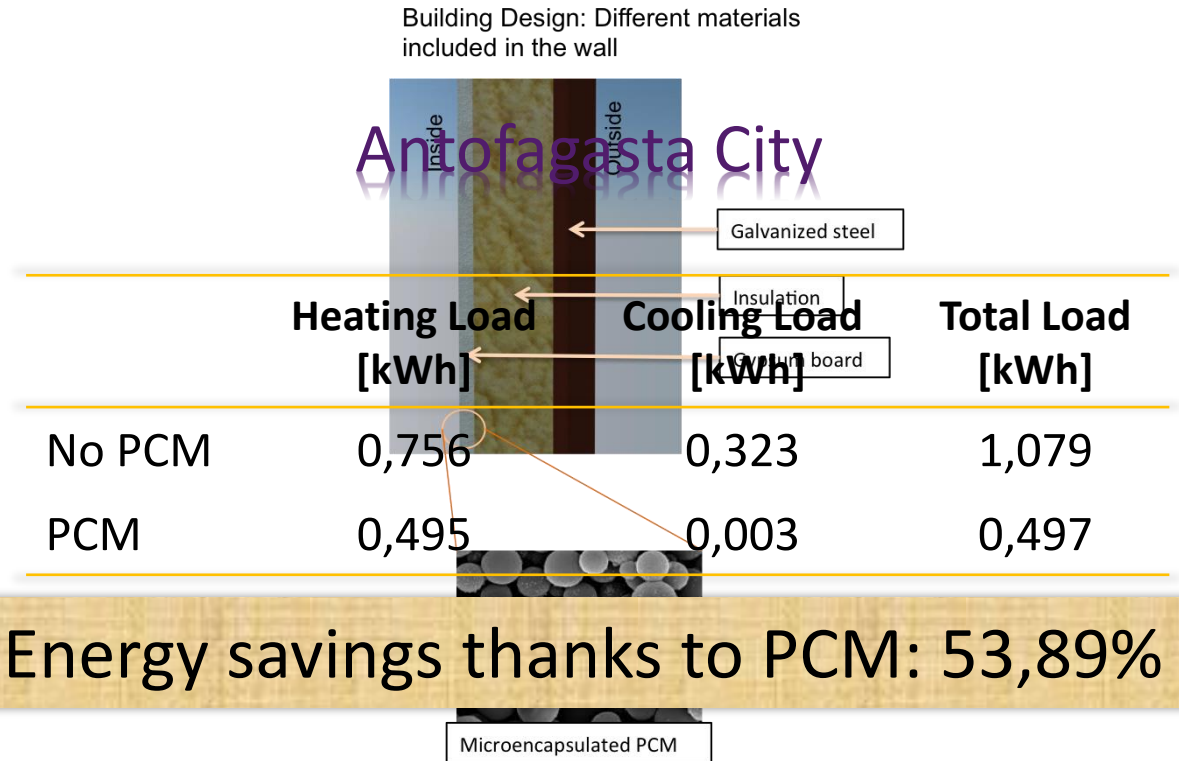
Energy Plus Simulations



Considering a theoretical geographical location:

- Antofagasta, Calama, Santiago and Concepción
- Containers:
 - with and without PCM
 - Distribution: north and south.
- Analysis of temperature fluctuations and energy consumption, for a period of one year.

The objective of this study was to evaluate the thermal and energy performance of containers of 2.5×2.5×2.5 m³.



Pilot-scale studies: Containers

Based on simulation results, containers had been constructed and sized with the same structural characteristics carried out in simulations.

Installation of sensors.


Installation of containers


Calibration of sensors

R.T.	M.V.	C.M.	U
0.000	-6.110	0.110	+0.005
-24.519	25.100	0.260	+0.006
25.681	25.000	0.071	+0.011
50.041	49.990	0.355	+0.011
75.051	74.966	0.360	+0.042
100.051	99.950	0.262	+0.037
125.050	124.870	0.160	+0.006
0.000	-6.110	0.110	-0.016


Collaborations

Universidad de Antofagasta 

Solar Energy Research Center, SERC Chile. 

National Commission for Scientific and Technological Research (CONICYT) 

CELiMIN  CENTRO DE INVESTIGACIÓN AVANZADA DEL LITIO Y MINERALES INDUSTRIALES

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Dr Mohammad Saffari



University of Auckland, New Zealand

Dr Mohammed Farid
MSc. Ryan Zhu





Overview

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Research proposal

General objective

Determine the feasibility of applying a TCS reactor systems using water as sorbent together with salts of different nature as sorbent (*i.e.* commercial salts and residual salts from mining industry) and redesign the store/heat exchanger to reduce the thermal mass to obtain a better heat transfer yield.



Research proposal

- **Specifics objectives**

Residual salts
Bischofite ($\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$)
Lithium carnallite ($\text{LiMgCl}_4 \cdot 6\text{H}_2\text{O}$)
Potassium carnallite ($\text{KMgCl}_4 \cdot 6\text{H}_2\text{O}$)
Kainite ($\text{KMg}(\text{SO}_4)\text{Cl} \cdot 3\text{H}_2\text{O}$)
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Theoretical evaluation of thermochemical sorption heat storage behavior of salts



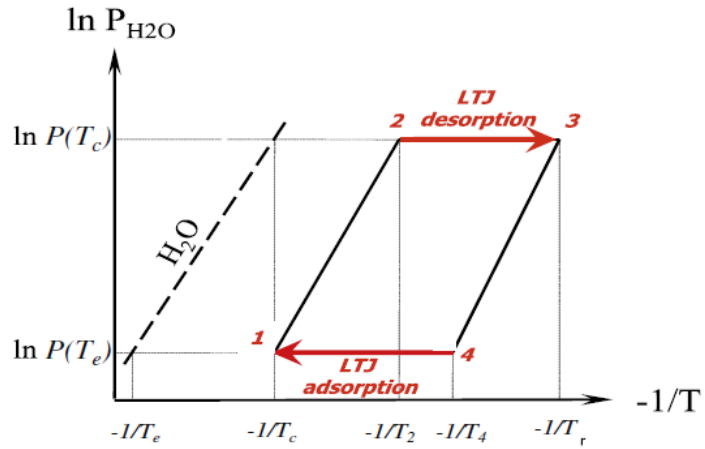
Salts characterisation



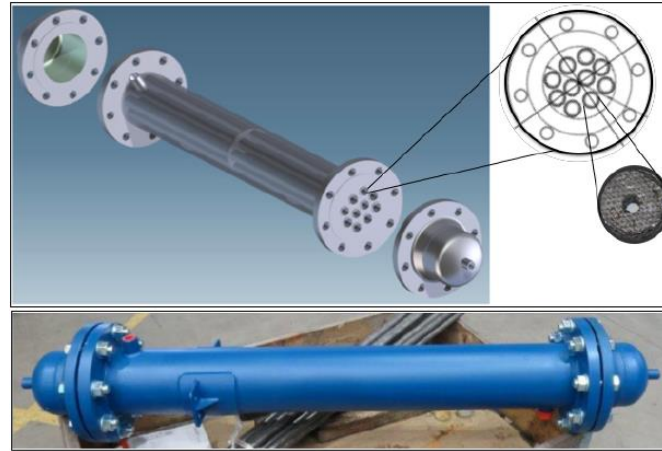
Preparation of composite materials

Research proposal

- **Specifics objectives**



Reaction rate test for the composite materials



Heat exchanger design



Heat exchanger tests

Work plan (Gantt chart)

Tasks	Year 1				Year 2			
Review of published articles								
Search for salts with appropriate reactions for application into TCS systems	×							
Prepare list of selected salts	×							
Determination of the energy storage capacity of salts. Chemical and Physical characterisation								
Chemical analysis of composition AA		×						
Identification of crystalline phases XRD		×						
Morphological analysis SEM		×						
Dehydration of samples TGA-DSC		×						
Thermal behaviour of the samples DSC		×						
Thermal stability and dynamic dehydration STA		×						
Determination of released gases and formation of solids TG-MS and XRD			×					
Determination of temperature limit conditions TG-MS and XRD			×					
Determination of the different stages of dehydration TG-MS and XRD			×					
Analysis of gaseous products MS			×					
Characterisation of thermal decomposition HT-XRD			×					
Preparation of composite matrix (ENG+salt)								
Conduct studies of conductivity, porosity, density and permeability to the ENG matrix				×				
Determine the best configuration of the ENG matrix					×			
Incorporate the sorbent (salt) into the ENG matrix					×			
Evaluation of dynamic reactions to the composite matrix using "large temperature jump" test								
Heat exchanger design								
Determination of design parameters						×		
Construction of the heat exchanger						×	×	
Test the heat exchanger (energy store) in the ThermExS lab facility								

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